## The Backbone Rearrangement of Cholesterol: a Chemical Proof

By P. BOURGUIGNON, J. C. JACQUESY, R. JACQUESY, J. LEVISALLES,\* and J. WAGNON

(Laboratoire de Chimie Organique, Faculté des Sciences de Nancy, France; Equipe de Recherches Associée au C.N.R.S.)

Summary Anhydrous HF transforms cholesterol into the two epimeric  $17\beta$ -hydroxy-3-methyl-3-(4-fluoro-4-methyl)pent-1-yl-D-homo-5 $\alpha$ -androstanes, through a backbone rearrangement followed by ring-expansion and hydride transfer.

BESIDES the expected addition products,<sup>1,2</sup> reaction of anhydrous HF with cholesterol (1) gives other compounds.<sup>3</sup> We report the structures of two of these, (A), m.p. 144°, and (E), m.p. 157—158°, isolated in 2% yields.

(A) and (E) are isomers,  $C_{27}H_{47}OF$ . <sup>1</sup>H and <sup>19</sup>F n.m.r. data show that they both contain the fragment  $CH_2 \cdot CFMe_2$  (methyl protons: doublet at  $\tau$  8.65; J 21 Hz; fluorine:

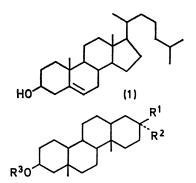
nonuplet; J 21 Hz). Spectroscopic arguments, which are omitted for the sake of brevity, lead to tentative structures (2) for (A) and (E). Both structures were confirmed by the synthesis of the degradation products (3).

Base-catalysed elimination of HF gave a mixture of olefins (4) and (5). NaIO<sub>4</sub>-KMnO<sub>4</sub> oxidation<sup>4</sup> of the mixture afforded, *inter alia*, acid (6), which was submitted to two successive Barbier-Wieland degradations, leading to acid (7) [(7A) from (A), (7E) from (E)]. Physical arguments indicated that (7A) had an axial carboxyl group, (7E) an equatorial carboxyl group. The corresponding esters (3A) (m.p. 120–122°) and (3E) (m.p. 121.5°) were synthesized in following way.

The mixture of the known<sup>5</sup> alcohols (8a) and (8b) was converted, by treatment with H2SO4-HCO2H,6 into acid (8c). The corresponding methyl ester (8d) was then submitted to a D-ring expansion by the usual treatment. Ketone (9A) was obtained as a minor product.  $NaBH_4$ reduction of (9A), followed by acetylation, gave a compound identical in every respect with (3A).

 $17\beta$ -Acetoxy-5 $\alpha$ -androstan-3-one<sup>7</sup> was transformed into nitrile (10a) by standard procedures. The sequence  $(10a) \rightarrow (10d)$  is straightforward. Methylenation  $(CH_2I_2 -$ Zn)<sup>8</sup> of alcohol (10d) gave alcohol (11a), whose acetate (11b) was hydrogenolysed<sup>9</sup> to acetate (12a). The corresponding diol (12b) was oxidized to acid (8e), whose ester (8f) was submitted as above to ring-D expansion. NaBH<sub>4</sub> reduction of the minor product (9E), followed by acetylation, gave a compound identical in every respect with (3E).

These results fully confirm formulae (2A) and (2E), and

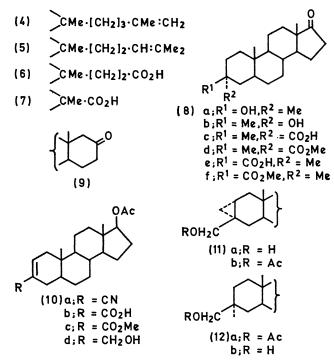


- (2A)  $R^1 = Me_1 R^2 = [CH_2]_3 \cdot CFMe_2 = R^3 = H$
- (2E)  $R^1 = [CH_2]_3 \cdot CFMe_2; R^2 = Me; R^3 = H$

(3A)  $R^1 = Me; R^2 = CO_2Me; R^3 = Ac$ 

(3E)  $R^1 = CO_2 Me_1 R^2 = Me_1 R^3 = Ac$ 

imply that this acid-catalysed rearrangement of cholesterol (1) is one of the most extensive known to date and goes much further than the usual backbone rearrangements of steroids.10-12



We thank C.N.R.S. for financial support and the Schering Corporation for gift of starting materials.

(Received, January 29th, 1970; Com. 136.)

<sup>†</sup>Cyanohydrin formation, reduction to amino-alcohol, and nitrous acid deamination.

- <sup>1</sup> C. S. Barnes and C. Djerassi, J. Amer. Chem. Soc., 1962, 84, 1962.
- <sup>2</sup> R. Jacquesy and J. Levisalles, Bull. Soc. chim. France, 1966, 1884.
- <sup>3</sup> J. C. Jacquesy, R. Jacquesy, and J. Levisalles, Bull. Soc. chim. France, 1967, 1649. <sup>4</sup> R. U. Lemieux and E. von Rudloff, Canad. J. Chem., 1965, 33, 1701 and 1710.
- <sup>5</sup> H. G. Lehmann, O. Engelfried, and R. Wiechert, J. Medicin. chem., 1965, 8, 383.

- <sup>6</sup> H. Koch and W. Haaf, Annalen, 1958, 618, 321.
  <sup>7</sup> F.L. Weisenborn and H. E. Applegate, J. Amer. Chem. Soc., 1959, 81, 1960.
  <sup>8</sup> H. E. Simmons and R. D. Smith, J. Amer. Chem. Soc., 1959, 81, 4256.
  <sup>9</sup> C. W. Woodworth, V. Buss, and P. von R. Schleyer, Chem. Comm., 1968, 569.
- <sup>10</sup> J. W. Blunt, M. P. Hartshorn, and D. N. Kirk, *Tetrahedron Letters*, 1966, 2175. <sup>11</sup> J. C. Jacquesy, J. Levisalles, and J. Wagnon, *Chem. Comm.*, 1967, 25.
- <sup>12</sup> F. Frappier, Q. K. Huu, F. X. Jarreau, J. Hannart, and R. Goutarel, Compt. rend., 1967, 264, C, 707.